



## “INVESTIGATION OF ACTIVE PHYTOCONSTITUENTS OF LEAVES BIXA ORELLANA EXTRACTS”

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### ABSTRACT

The plant *Bixa orellana* Linn., its affined to family Bixaceae. Is a small evergreen shrub or bushy tree commonly known “annatto” in English and “Sinduri” in Sanskrit<sup>[1]</sup>, The prickly heart shaped pods contain small reddish-orange seeds. each pod contains approximately 50 seeds. That is native to central, south and north America. This is concluded that methanolic extract and aqueous extract are more prone to positive result rather than oil extract. So that *Bixa orellana* will show their positive activity on aqueous and alcoholic extract and many constituet found positive. The crude extract of *Bixa orellana* leaves was subjected to a preliminary phytochemical screening for the presence of steroids,

flavonoids, tannins, saponins, alkaloids, glycoside, cumerin, and carbohydrates. Carcinogenesis is a multistep process, and oxidative damage is linked to formation of tumors through several mechanisms. Oxidative stresses induced by free radicals and cause DNA damage, which, when left unrepaired, can lead to base mutation, single and double strand unseat, DNA cross-linking, and chromosomal damage and rearrangement.

**KEYWORDS:** *Bixa orellana*, tannins, sindur, methanol.

### INTRODUCTION

The plant *Bixa orellana* Linn., its affined to family Bixaceae. Is a small evergreen shrub or bushy tree commonly known “annatto” in English and “Sinduri” in Sanskrit<sup>[1]</sup>, The prickly heart shaped pods contain small reddish-orange seeds. each pod contains approximately 50 seeds. That is native to central, south and north America. It is also known as achiote or annatto. and it is found mainly in the forest ecosystem of the Brazilian humid tropics but it is now cultivated in the tropics throughout the world.<sup>[3]</sup> In addition, it enjoys folkloric use in

india, Bangladesh, china and the Philippines. The bixa orellana red seeds serve as the source of a widely used. Red to yellow dye which is referred to as annatto and contains primarily the carotenoids bixin and norbixin. Red dye is used to color and clothes, and various food products including Rice, butter, cheeses, soups, and soft drinks. and make Red body paint, and lipstick, as well as a spice.<sup>[5]</sup> For this reason, the achiote is sometimes called the lipstick tree. It is many medicinal uses in traditional system of medicine including Ayurveda. Root and leaves are useful in the treatment of sore throat, jaundice, liver diseases, diabetes and hypertension.

### ***(PLANT PROFILE)***

#### ***(Bixa orellana)***

Kingdom : Plantae

Clade : Angiosperms

Clade : Rosids

Order : Malvales

Family : Bixaceae

Species : *B. orellana*

Binomial name : *Bixa orellana* L.

### ***SYNONYMS***

- *Bixa tinctoria* Salisb.
- *Orellana orellana* (L.) Kuntze
- *Bixa americana* Poir.



**Plant of BIXA ORELLANA**

## MATERIALS AND METHODS

### 1. Chemical and reagents

Methanol was obtained from “AVANTOR PERFORMANCE MATERIAL MAHARASHTRA INDIA LIMITED” and distilled water was obtained from “NRI INSTITUTE LABORATORY BHOPAL” and vegetable oil (soyabin oil) was obtained from.

### 2. Plant material and collection

The leaves of *Bixa orellana* were collected from the plant grown in the botanical garden of department of “NRI INSTITUTE OF PHARMACEUTICAL SCIENCES BHOPAL” after identification by a guide. A voucher specimen is deposited in the laboratory herbarium. The collected plant material was thoroughly checked and free from foreign matter and shade dried. Powdered with laboratory mixer and sieved.

### 3. Mixing

After shade dried leaf are powder through help of mixture in laboratory.



### 4. Sieving

After mixing leaf powder are sieving from #22 no sieves. And after sieving make sure powder does not contain any row material. if should be contain. than clean the sample.

## 5. Prepration of Extract

### (a). soxhlet method

The weighed (50g) leaf powder was kept to successive Soxhlet extraction using methanol and water (60-80°C), the solvent should be taken 3/4th of the round bottomed flask, each for 6 hours and each solvent should be extracted single time. The obtained solvent extracts were concentrated and dried in desiccators.



### (b). decoction method

The weight acuratly (10g) leaf powder was keep for decoction extraction by using vegetable oil (soyabin oil) at room temprature. The solvent should be taken 3/4<sup>th</sup> of the beackar. for 72 hours. And solvent should be extract after 72 hours. The obtained solvent extract and perform phytochemical test.



## PRELIMINARY PHYTOCHEMICAL SCREENING

The crude extract of *Bixa orellana* leaves was subjected to a preliminary phytochemical screening for the presence of steroids, flavonoids, tannins, saponins, alkaloids, glycoside, cumerin, and carbohydrates.

### 1. TESTS FOR CARBOHYDRATES

**1. Molisch's Test** (general test):- To 2-3 ml aqueous extract, added few drops of alpha-naphthol solution in alcohol, shake and add conc. H<sub>2</sub>SO<sub>4</sub> from side of the test tube. Violet ring is formed at the junction of two liquid.

#### (A). TESTS FOR REDUCING SUGARS

**(a). Fehling's Test:-** Mix 1 ml of "Fehling's A" and "Fehling's B" solution boil for one minute. Add equal volume of test solution. Heat in boiling water bath for 5-10 minute. First a yellow, than brick red precipitate is observed.

**(b). Benedict's Test:-** Mix equal volume of Benedict's reagent and test solution in test tube. Heat in boiling water bath for 5 min. solution appears green, yellow or Red depending of amount of reducing sugar present in test solution.

#### (B). TEST FOR MONOSACCHARIDE:-

**(a) Barfoed's test:-** Mix equal volume of Barfoed's reagent and test solution. Heat for 1-2 min. in boiling water bath. And cool Red precipitation is observed.

#### (C). TEST FOR PENTOSE SUGARS:-

**(a). Bial's test:-** To boiling Bial's reagent add few drops of test solution. green or purple coloration appears.

#### (D). TESTS FOR HEXOSE SUGARS

**(a). Seliwanoff's test ( for keto hexose like fructose):-** Heat 3 ml Seliwanoff's reagent and 1 ml.

Test solution in heating water bath for 1-2 minute Red color is formed.

**(b). Tollens' Test (phloroglucinol test for galactose):-** Mix 2.5 ml conc HCL and 4 ml 0.5% phloroglucinol add 1-2 ml. test solution. Heat yellow to Red color appears.

### 2. TESTS FOR STEROIDS

**(a). Salowski reaction:-** To 2 ml of extract, add 2 ml. of chloroform, and 2 ml. conc. H<sub>2</sub>SO<sub>4</sub> shake well, chloroform layer appears Red and acid layer shows greenish yellow fluorescence.

(b) **Libermann – Burchard reaction:-** Mix 2 ml. extract with chloroform add 1-2 ml. acetic – anhydride and 2 drops conc. H<sub>2</sub>SO<sub>4</sub> from the side of test tube. First Red, then blue and finally green color appears.

(c). **Liebermann's reaction:-** Mix 3 ml. extract with 3 ml of acetic anhydride. Heat and Cool. And few drops of conc. H<sub>2</sub>SO<sub>4</sub>. Blue color appears.

### 3. TESTS FOR ALKALOIDS

Evaporate the aqueous, alcoholic and chloroform extracts separately, To residue, add dilute HCL shake well and filter with filtrate perform following test.

(a). **Dragendorff's test:-** To 2-3 ml of filtrate, add few drops Dragendorff's reagent. Orange brown precipitate formed.

(b). **hager's test:-** 2-3 ml. of filtrate with hager's reagent gives yellow precipitate.

### 4. TESTS FOR GLYCOSIDES

Determine free sugar content of extract, hydrolyse the extract with mineral acid (dilute HCL/ dilute H<sub>2</sub>SO<sub>4</sub>). Again determine the total sugar content of the hydrolysed extract. Increase the sugar content indicates presence of glycoside in the extract.

#### (A). TESTS FOR CARDIAC GLYCOSIDES

(a). **Baljet test:-** Take a piece of lamina or thick section of the leaf and add sodium picrate reagent. If glycoside is present yellow to orange colour will be seen.

(b). **Legal test:-** To aqueous or alcoholic extract, add 1 ml of pyridine and 1 drop of nitroprusside. Pink to Red colour is appears.

(c). **Killer killiani test (test for deoxysugars):-** To 2 ml. extract added glacial acetic acid. One drop Of 5% FeCl<sub>3</sub> and conc. H<sub>2</sub>SO<sub>4</sub>. Raddish brown color appears at junction of the two liquid layer and upper layer appears blueish green.

#### (B). TESTS FOR ANTHRAQUINONE GLYCOSIDES

(a). **Borntranger's test:-** To 3 ml of extract, add dilute H<sub>2</sub>SO<sub>4</sub>. Boil and filter. To cold filtrate add equal volume benzene or chloroform. Shake well. Separate the organic solvent add ammonia. Ammoniacal layer turns pink or Red.

(b). **Modified Anthraquinones test:-** To 5 ml of extract, add 5ml. 5% FeCl<sub>3</sub> and 5 ml diluted HCL Heat for 5 minute in boiling water bath. Cool and add benzene or any organic solvent shake well. Separate organic layer add equal dilute ammonia. Ammoniacal layer shows pinkish Red color.

**(C). TESTS FOR SAPONIN GLYCOSIDE**

(a). **Foam test:-** shake the extract or dry powder vigorously with water persistent foam observed.

(b). **Haemolytic test:-** Add drug extract or dry powder to one drop of blood place on glass slide. “ Haemolytic zone appears.”

**(D). TEST FOR FLAVONOIDS**

**Shinoda test:-** (a). To dry powder or extract, add 5 ml. of 95% ethanol, and few drop of conc. HCL and 0.5g magnesium turnings, pink color observed.

(b). To small quantity of residue, add lead acetate solution yellow colored precipitate is formed.

(c). Addition of increasing amount of sodium hydroxide to the residue shows yellow coloration, which decoloration after addition of acid.

**5. TESTS FOR TANNINS AND PHENOLIC COMPOUND**

To 2-3 ml of aqueous or alcoholic extract, add few drop of following reagents.

(a). **5% FeCl<sub>3</sub> solution:-** deep blue-black color.

(b). **Lead acetate solution:-** white precipitate

(c). **Gelatin solution:-** white precipitate.

(d). **Bromine water:-** decoloration of bromine water.

(e). **Acetic acid solution:-** Red color solution.

(f). **Potassium dichromate:-** Red precipitate.

(g). **Dilute iodine solution:-** Transient Red color.

(h). **Dilute potassium permanganate solution:-** Decoloration.

**RESULT AND DISCUSSION****1. TEST FOR CARBOHYDRATES**

(a). **Molisch's Test:-**

S. No	Solvent	Inference
1.	Methanol	- ve
2.	Water	+ ve
3.	Oil	- ve

**(b). Fehling's test(for reducing sugar)**

S. No	Solvent	Inference
1.	Methanol	- ve
2.	Water	+ ve
3.	Oil	- ve

**(C). Benedict's test**

S. No	Solvent	Inference
1.	Methanol	+ ve
2.	Water	+ ve
3.	Oil	- ve

**(d). Barfoed's test( for monosaccharide sugar)**

S. No	Solvent	Inference
1.	Methanol	- ve
2.	Water	+ ve
3.	Oil	- ve

**(e). Bial's test( for pentose sugar)**

S. No	Solvent	Inference
1.	Methanol	+ ve
2.	Water	- ve
3.	Oil	- ve

**(f). Seliwanoff's test( for hexose sugar):-**

S. No	Solvent	Inference
1.	Methanol	- ve
2.	Water	+ ve
3.	Oil	- ve

**(g). Tollens' Test**

S. No	Solvent	Inference
1.	Methanol	- ve
2.	Water	+ ve
3.	Oil	- ve

**(2). TESTS FOR GLYCOSIDES****(a). Legal test(for cardiac glycoside)**

S. No	Solvent	Inference
1.	Methanol	- ve
2.	Water	- ve
3.	Oil	- ve



**(b). Killer killiani test (test for deoxysugars)**

S. No	Solvent	Inference
1.	Methanol	- ve
2.	Water	- ve
3.	Oil	+ ve

**(c). Borntranger's test( for anthraquinone glycoside)**

S. No	Solvent	Inference
1.	Methanol	+ Ve
2.	Water	- Ve
3.	Oil	- Ve

**(d). Modified Borntranger's test**

S. No	Solvent	Inference
1.	Methanol	+ ve
2.	Water	- ve
3.	Oil	- ve

**(e). Foam test( for saponine glycoside)**

S. No	Solvent	Inference
1.	Methanol	- ve
2.	Water	- ve
3.	Oil	- ve

**(f). Haemolytic test**

S. No	Solvent	Inference
1.	Methanol	+ve
2.	Water	- ve
3.	Oil	+ ve

**(3). TESTS FOR ALKALOIDS****(a). Dragendorff's test**

S. No	Solvent	Inference
1.	Methanol	+ve
2.	Water	- ve
3.	Oil	- ve

**(b). Wagner's test**

S. No	Solvent	Inference
1.	Methanol	- ve
2.	Water	- ve
3.	Oil	- ve

## (c). Hager's test

S. No	Solvent	Inference
1.	Methanol	+ ve
2.	Water	+ ve
3.	Oil	- ve

## (4). TESTS FOR STEROIDS

## (a). Salowski test(reaction)

S. No	Solvent	Inference
1.	Methanol	- ve
2.	Water	- ve
3.	Oil	- ve

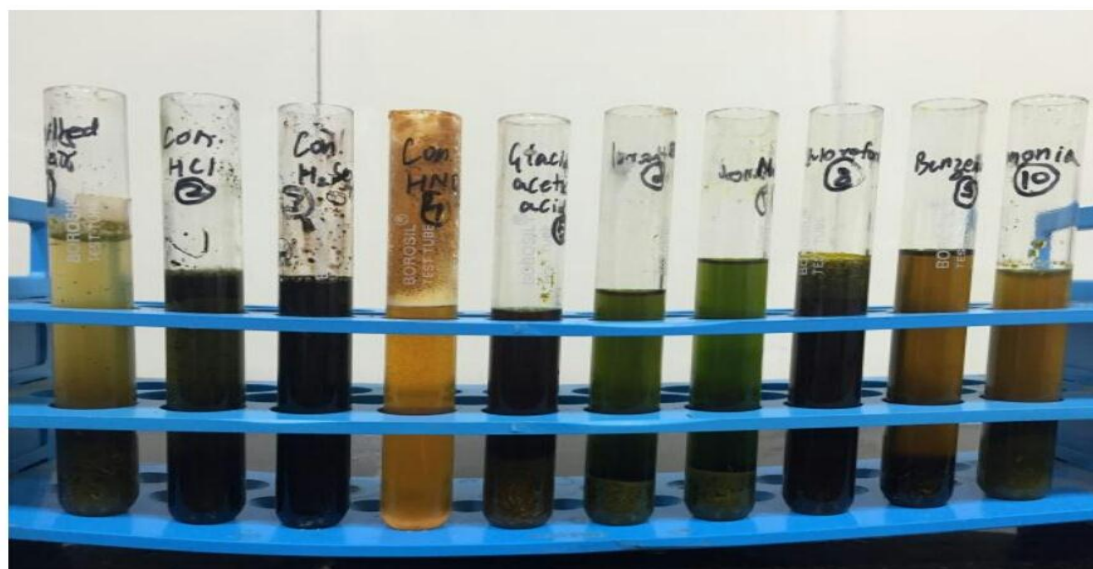
## (b). Libermann – Burchard reaction

S. No	Solvent	Inference
1.	Methanol	+ ve
2.	Water	- ve
3.	Oil	- ve

## 5. TEST FOR FLAVONOIDS

## (a). Shinoda test

S. No	Solvent	Step 1	Step 2	Step 3	Step 4
1.	Methanol	- ve	+ ve	- ve	- ve
2.	Water	- ve	+ ve	+ ve	+ ve
3.	Oil	- ve	- ve	+ ve	+ ve



**6. TESTS FOR TANNINS AND PHENOLIC COMPOUNDS**

S. No	Test (reagents)	Methanol	Water	Oil
1.	<b>5% FeCl<sub>3</sub> solution</b>	- ve	+ ve	- ve
2.	<b>Lead acetate solution</b>	- ve	+ ve	- ve
3.	<b>Gelatin solution</b>	+ ve	- ve	- ve
4.	<b>Bromine water</b>	ve	Ve	ve
5.	<b>Acetic acid solution</b>	- ve	- ve	- ve
6.	<b>Potassium dichromate</b>	- ve	- ve	- ve
7.	<b>Dilute Iodine solution</b>	- ve	+ ve	- ve
8.	<b>Dilute HNO<sub>3</sub></b>	- ve	+ ve	- ve
9.	<b>Dilute potassium permanganate solution</b>	- ve	- ve	- ve

**CONCLUSION**

This is concluded that methanolic extract and aqueous extract are more prone to positive result rather than oil extract. So that Bixa orellana will show their positive activity on aqueous and alcoholic extract and many constituent found positive.

Deoxy sugar show the presence in oil extract and hemolytic test is also positive, flavanoid is soluble in oil.

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